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SYNTHESIS AND CHARACTERIZATION OF Na AND Mg CONTAINING CARBONATEFLUORAPATITES

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The apatites were precipitated in an aqueous NH₄OH-NH₄NO₃ solution with pH 9-10 at temperatures 20°C and 80°C. The synthesized materials were studied by chemical and thermal analyses (TG/DTG/DTA, TG/FTIR-EGA), IR-spectroscopy, XRD-powder analysis and specific area measurements.

The precipitated materials are poorly crystallized carbonatefluorapatites (CFAp) with a large specific area. Their chemical composition can be expressed with the general formula $Ca_aMg_bX_c(HPO_4)_x(PO_4)_{6-x-d}(CO_3)_dF_{2-y}(OH)_y$ nH_2O , where X is Na⁺ or NH₄⁺ and a=4-10, b=0-5, c(Na⁺)=0,2-0,4, c(NH₄⁺)=0-0,9 and d=0,05-0,75. A higher temperature of precipitation diminishes the content of CO_3^{2-} and CO_3^{2-} and CO

On heating water evolves stepwise at the temperatures up to 500°C and ammonia in the temperature range 100-400°C, followed by the appearance of P-O-P bonds. Carbon dioxide evolves in a wide temperature range from 100°C up to 800-1000°C. The heating products consist of well-crystallized CO₃²⁻ free apatite.